

Introduction to Chemical Reactors

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6. Non-ideal reactors

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Motivation

PFR (completely unmixed) and CSTR (completely mixed) reactors are never achieved in practice, however, the behaviour of real reactors can be modelled as a combination of both.

Interestingly:

- a PFR can be modelled as an infinite number of CSTR in series
- a CSTR can be modelled as a PFR with infinite recycle

Any real reactor can be conceptualised as a collection of small elements each one with an associated residence time.

The concentration in each element is a function of its residence time.

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Objectives

- Introduce the concepts of
 - Residence Time Distribution (RTD)
 - mean residence time
 - mean outlet concentration
 - mean conversion
- Apply RTD to calculate the concentration and conversion in the stream exiting a reactor.

Sources for non-ideal reactors

- Metcalfe: Chapter 6
- Schmidt: Chapter 8
- Fogler: Chapters 13 and 14

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RTD (Residence Time Distribution)

RTD (Residence Time Distribution) is the probability of an element of fluid residing in the reactor for a time t .

RTDs, denoted $E(t)$, can be obtained by injecting a tracer at the input stream and then monitoring its concentration in the outlet.

$$E(t) \equiv \frac{C(t)}{\int_0^\infty C(t) dt}$$

normalised

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RTD for an ideal CSTR

Introduce N moles of tracer at $t = 0$:

$$C_{init} = \frac{N}{V}$$

Non-SS CSTR MB: $V \frac{dC}{dt} = v(C_0 - C) + v_f r$

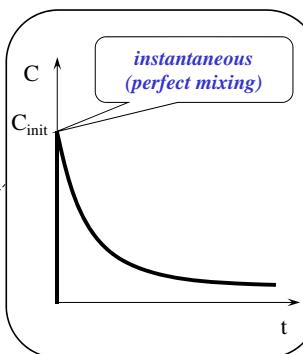
$$\tau \frac{dC}{dt} = -C \quad BC: C = C_{init} @ t = 0$$

$$\text{solution: } C(t) = C_{init} e^{-\frac{t}{\tau}}$$

$$E(t) \equiv \frac{C(t)}{\int_0^\infty C(t) dt} = \frac{C_{init} e^{-\frac{t}{\tau}}}{C_{init} \tau} \longrightarrow E(t)_{CSTR} = \frac{1}{\tau} e^{-\frac{t}{\tau}}$$

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RTD for an ideal PFR

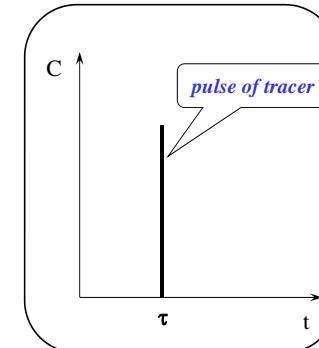
Introduce a pulse of N moles of tracer at $t = 0$.

$$E(t)_{PFR} = \delta(t - \tau)$$

where $\delta(t - \tau) = 0$ for $t \neq \tau$

$\delta(t - \tau) \neq 0$ for $t = \tau$

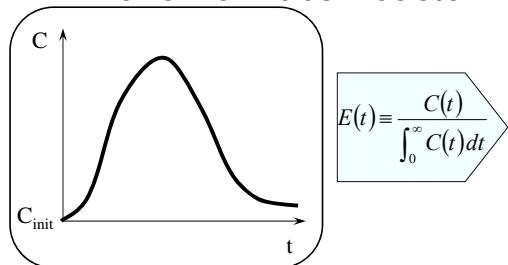
$$\int \delta(t - \tau) dt = 1$$



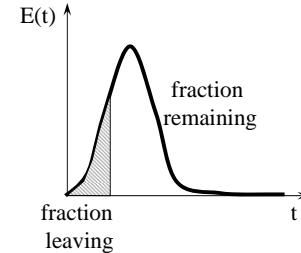
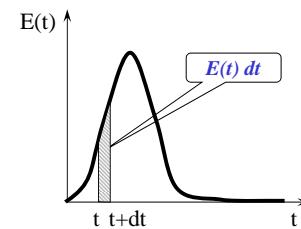
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RTD for a non-ideal reactor



$E(t) dt$: fraction of effluent with a residence time between t and $(t+dt)$



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Mean residence time (\bar{t})

\bar{t} is the time-weighted average over all residence times:

$$\bar{t} = \int_0^\infty t E(t) dt$$

for ideal reactors:

$$\bar{t}_{CSTR} = \int_0^\infty t \left[\frac{1}{\tau} e^{-\frac{t}{\tau}} \right] dt = \left[-t e^{-\frac{t}{\tau}} \right]_0^\infty + \int_0^\infty e^{-\frac{t}{\tau}} dt = \tau$$

by parts

$$\bar{t}_{PFR} = \int_0^\infty t \delta(t - \tau) dt = \tau$$

δ(t-τ) is zero for all times except t = τ

as expected!

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Average values

In fact, the average value of any variable $f(t)$ is:

$$\bar{f} \equiv \int_0^\infty f(t) \cdot E(t) dt$$

The **mean outlet concentration** of reactant:

$$\bar{C}_A \equiv \int_0^\infty C_A(t) E(t) dt$$

The **mean conversion** in a non-ideal reactor:

$$\bar{X} \equiv \int_0^\infty X(t) \cdot E(t) dt$$

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Mean conversion (\bar{X})

If we treat each fluid element in the reactor as a well-mixed batch reactor (for a first order reaction A \rightarrow B):

$$C_A = C_{A0} \exp[-kt] \quad \rightarrow \quad \frac{C_A}{C_{A0}} = 1 - X = \exp[-kt] \quad \rightarrow \quad X = 1 - \exp[-kt]$$

for ideal reactors:

$$\bar{X}_{CSTR} = \int_0^\infty (1 - e^{-kt}) \left[\frac{1}{\tau} e^{-t/\tau} \right] dt = \left[\frac{1}{\tau} (-\tau) e^{-t/\tau} \right]_0^\infty - \frac{1}{\tau} \int_0^\infty e^{-kt-\frac{t}{\tau}} dt = \frac{k\tau}{1+k\tau}$$

by parts

$$\bar{X}_{PFR} = \int_0^\infty (1 - e^{-kt}) \cdot \delta(t - \tau) dt = 1 - e^{-k\tau}$$

$\delta(t-\tau)$ is zero for all times except $t = \tau$

again, as expected!

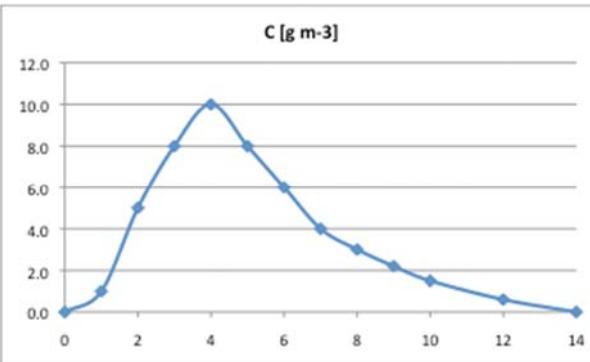
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Example: C(t) curve for tracers (Example 13-1 in Fogler)

A tracer is injected as a pulse to a reactor and the effluent concentration is measured:

t [min]	C [g m ⁻³]
0	0.0
1	1.0
2	5.0
3	8.0
4	10.0
5	8.0
6	6.0
7	4.0
8	3.0
9	2.2
10	1.5
12	0.6
14	0.0



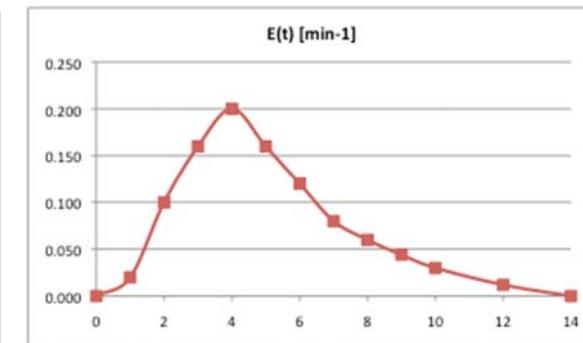
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Example: E(t) curve for tracers

$$E(t) \equiv \frac{C(t)}{\int_0^\infty C(t) dt} = \frac{C(t)}{50.033} \quad (\text{the integral is evaluated numerically using Simpson's rule})$$

t [min]	$E(t)$ [min ⁻¹]
0	0.000
1	0.020
2	0.100
3	0.160
4	0.200
5	0.160
6	0.120
7	0.080
8	0.060
9	0.044
10	0.030
12	0.012
14	0.000



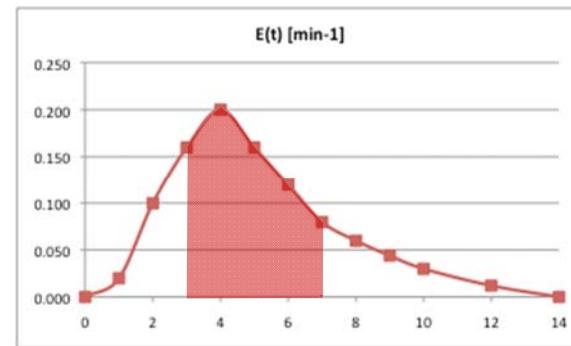
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Example: fraction of material with a given residence time

What is the fraction of the material that has spent between 3 and 7 minutes in the reactor?

t [min]	E(t) [min-1]
0	0.000
1	0.020
2	0.100
3	0.160
4	0.200
5	0.160
6	0.120
7	0.080
8	0.060
9	0.044
10	0.030
12	0.012
14	0.000



$$\int_3^7 E(t) dt = \frac{1}{3} [0.16 + 4(0.2) + 2(0.16) + 4(0.12) + 0.08] = 0.61$$

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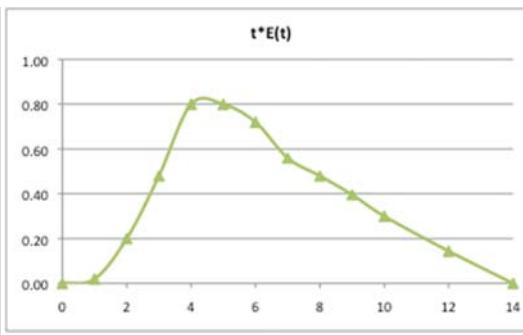
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Example: mean residence time for tracers

$$\bar{t} = \int_0^\infty t E(t) dt = 5.155 \text{ min}$$

(t_{mean} is the mean residence time; it is also evaluated using Simpson's rule)

t [min]	E(t) [min-1]	t * E(t)
0	0,000	0,00
1	0,020	0,02
2	0,100	0,20
3	0,160	0,48
4	0,200	0,80
5	0,160	0,80
6	0,120	0,72
7	0,080	0,56
8	0,060	0,48
9	0,044	0,40
10	0,030	0,30
12	0,012	0,14
14	0,000	0,00



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(Example: numerical integration)

For $N+1$ points (N even):

$$\int_{X_0}^{X_N} f(X) dX = \frac{h}{3} [f_0 + 4f_1 + 2f_2 + 4f_3 + 2f_4 + \dots + 4f_{N-1} + f_N]$$

$$\text{where } h = \frac{X_N - X_0}{N}$$

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Laminar flow tubular reactors

If a reactor has mixed flow, $E(t)$ will be between the limits of a perfectly mixed (CSTR) and unmixed (PFR) reactors.

In a PFR:

- if u is low $\rightarrow Re < 2100 \rightarrow$ laminar flow (no radial or axial mixing)
- if u is high $\rightarrow Re > 2100 \rightarrow$ turbulent flow (axial dispersion)

For laminar flow, the velocity profile is:

$$u(r) = U_{\max} \left[1 - \left(\frac{r}{R} \right)^2 \right]$$

$$U_{\max} = 2\bar{u}$$

$$\bar{u} = \frac{v}{A}$$

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Laminar flow tubular reactors

$$E(t) \equiv \frac{\text{fraction of total fluid passing between } r \text{ and } (r+dr)}{\text{total fluid passing through the reactor}}$$

$$= \frac{\int_0^r (2\pi r) u(r) dr}{\int_0^R (2\pi r) u(r) dr} = \frac{\int_0^r (2\pi r) u(r) dr}{v} = \frac{\tau}{2t^3}$$

(complete derivation
@ Fogler, 3rd ed.,
Section 13.4.3)

So, the RTD for a laminar flow tubular reactor is:

$$E(t) = \frac{\tau}{2t^3}$$

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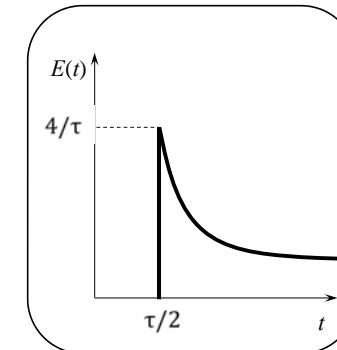
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Laminar flow tubular reactors: average conversion

Note that the minimum time that a fraction of fluid spends in the reactor (t_{\min}) is:

$$t_{\min} = \frac{L}{U_{\max}} = \frac{L}{2u} \left(\frac{\pi R^2}{\pi R^2} \right) = \frac{V}{2v} = \frac{\tau}{2}$$

$$\begin{aligned} \text{So, } E(t) &= 0 & t < \frac{\tau}{2} \\ E(t) &= \frac{\tau^2}{2t^3} & t \geq \frac{\tau}{2} \end{aligned}$$



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Laminar flow tubular reactors: average conversion

Since $\boxed{\bar{X} \equiv \int_0^\infty X(t) \cdot E(t) dt}$

$$\bar{X}_{\text{laminar flow PFR}} = \int_{\frac{\tau}{2}}^\infty X(t) \cdot E(t) dt = \int_{\frac{\tau}{2}}^\infty (1 - \exp[-kt]) \cdot \frac{\tau^2}{2t^3} dt$$

and, comparing the conversion of a laminar flow PFR vs a PFR for 1st and 2nd order kinetics:

$$\left. \frac{\bar{X}_{\text{laminar flow PFR}}}{\bar{X}_{\text{PFR}}} = \frac{\int_{\frac{\tau}{2}}^\infty (1 - \exp[-kt]) \cdot \frac{\tau^2}{2t^3} dt}{1 - \exp[-k\tau]} \right\} > 0.88 \text{ (worst case)}$$

$$= 1.0 \text{ (best case)}$$

... PFR assumption are a good approximation!

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Non-ideal reactors

We treat each fluid element in the reactor as a well-mixed batch reactor.

For a batch reactor (with a 1st order, irreversible reaction A → B):

$$\text{design equation: } \frac{dC_A}{dt} = v_j r = -kC_A \xrightarrow{\text{soln}} C_A = C_{A0} e^{-kt}$$

$$\boxed{\bar{C}_A = \int_0^\infty C_A(t) E(t) dt = C_{A0} \int_0^\infty e^{-kt} E(t) dt}$$

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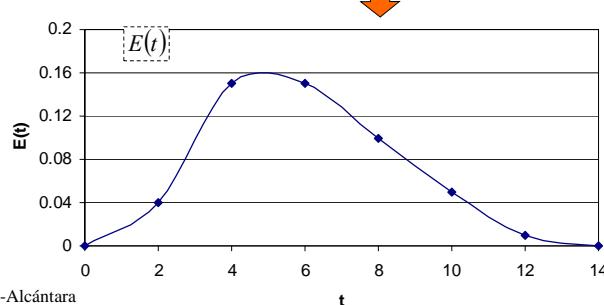
Non-ideal reactors: example (1) (6.2 Metcalfe)

Input data:

- first order, liquid phase reaction $A \rightarrow B$, $r = k C_A$, $k = 0.307 \text{ s}^{-1}$

t [s]	0	2	4	6	8	10	12	14
E [s ⁻¹]	0.00	0.04	0.15	0.15	0.10	0.05	0.01	0.00

- $E(t)$ vs. t

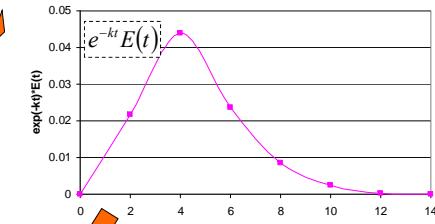
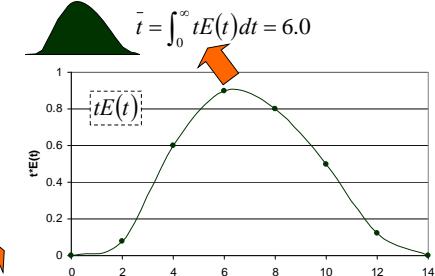
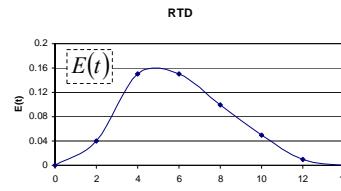


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Non-ideal reactors: example (2) (6.2 Metcalfe)

Input data:



$$\frac{C_A}{C_{A0}} = 1 - X_A = \int_0^{\infty} e^{-kt} E(t) dt = 0.201 \quad 6-22$$

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Non-ideal reactors: example (3) (6.2 Metcalfe)

Solution: $X_{A,\text{non-ideal}} = 1 - 0.201 = 0.799$

Compare with CSTR and PFR:

• CSTR $\frac{C_A}{C_{A0}} = \frac{1}{1+k\tau} \Rightarrow X_{A,\text{CSTR}} = 1 - \frac{1}{1+k\tau} = \frac{k\tau}{1+k\tau} = 0.65$

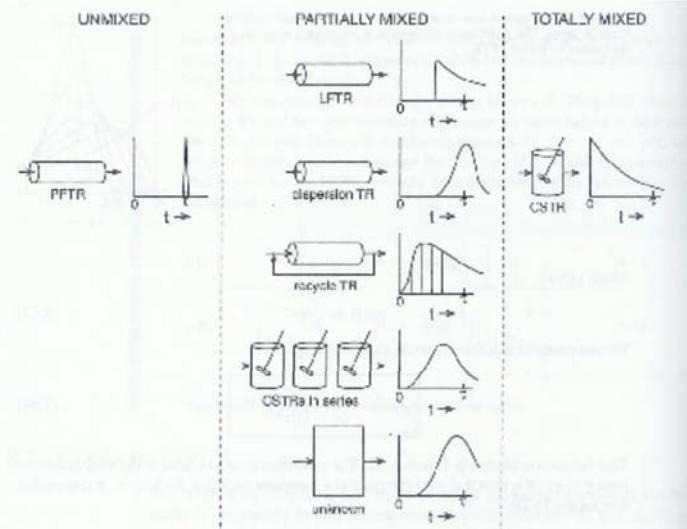
$$X_A = 1 - \frac{C_A}{C_{A0}} ; \quad \tau = 6 \text{ s}$$

• PFR $\frac{C_A}{C_{A0}} = \exp[-k\tau] \Rightarrow X_{A,\text{PFR}} = 1 - \exp[-k\tau] = 0.84$

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Some RTDs for non-ideal reactors (Fig 8-1, Schmidt)



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